

# Design, Synthesis, and Herbicidal Activities of Novel 2-Cyanoacrylates Containing Isoxazole Moieties<sup>†</sup>

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A series of novel 2-cyanoacrylates containing an isoxazole moiety were designed and synthesized. Their structures were characterized by <sup>1</sup>H NMR and elemental analysis (or high-resolution mass spectrometry). Their herbicidal activities against four species were evaluated, and the results indicated that some of the title compounds showed excellent herbicidal activities against rape and amaranth pigweed in postemergence treatment even at a dose of 75 g/ha.

KEYWORDS: Cyanoacrylates; isoxazole; herbicidal activity; PSII electron transfer inhibitor

## INTRODUCTION

2-Cyanoacrylates are inhibitors of photosystem II (PSII) electron transport, which inhibits the growth of weeds by disrupting photosynthetic electron transport at the PSII reaction center (1). Therefore, this kind of compound can be used as a potential herbicide. In previous works, we have reported the syntheses of cyanoacrylates containing a pyridine group (2,3), of which compound A gave excellent herbicidal activities especially against dicotyledon species. A series of heterocycles such as thiazole, furan, tetrahydrofuran, pyridazine, and pyrimidine groups were then introduced to cyanoacrylates (structure **B**), and the groups showed notable influence on the herbicidal activities of corresponding compounds (4-6) The quantitive structure-activity relationship research by comparative molecular field analysis (CoMFA) indicated that a bulky and electronegative group around the para position of proper aromatic rings would have the potential for higher activity (6). Moreover, the binding model of compound A with the D1 protein of PSII was built, and it was proved that the N atom on the pyridine ring could form an H-bond with the backbone amide of Phe265 on the D1 protein (7). Therefore, a series of new 2-cyanoacrylates bearing an N-containing heterocycle such as an oxazole, oxadiazole, or quinoline group (structure **B**) were designed, and some of them indeed exhibited good herbicidal activities (8). Among those compounds, the 3-phenyl-1,2,4-oxadiazol-5-yl moieties (structure C) seemed to have helped with their herbicidal activities, because most of them exhibited higher activities against all four tested species than oxazole-containing compounds (8). 1,2-Isoxazole, a structure very similar to 1,2,4-oxadiazole and also the important moiety in many activity compounds (9, 10), was then considered as a bioisosteric analogue to introduce the 2-cyanoacrylates. In this paper, we focus on the isoxazolecontaining 2-cyanoacrylates bearing different substituents at the 3-position of the 1,2-isoxazole ring (compound 1 and Table 1) and report the exciting results.



#### MATERIALS AND METHODS

**Instruments.** <sup>1</sup>H NMR spectra were obtained at 300 MHz using a Bruker AV300 spectrometer or at 400 MHz using a Varian Mercury Plus400 spectrometer in CDCl<sub>3</sub> solution with tetramethylsilane as the internal standard. Chemical shift values ( $\delta$ ) are given in parts per million. Elemental analyses were determined on a Yanaca CHN Corder MT-3 elemental analyzer. High-resolution mass spectrometry (HRMS) data were obtained on a FTICR-MS instrument (Ionspec 7.0T). The melting points were determined on an X-4 binocular microscope melting point apparatus (Beijing Technical Instruments Co., Beijing, China) and are uncorrected. Yields were not optimized.

Oximes 3a-3h were prepared from corresponding aldehyde according to a classical procedure (11). Dibromoformaloxime was prepared according to a published procedure (12). Compounds 7a and 7b were prepared according to our previous work (2, 4). All of the anhydrous solvents were dried and distilled by using standard techniques.

General Synthetic Procedure for 4a-4h (13). To a cooled (below 0 °C) solution of aldehyde oxime 3 (10 mmol), propargyl bromide (12 mmol), and triethylamine (10 mmol) in dichloromathane (20 mL) was dropwise added 8% aqueous sodium hypochlorite (30 mL). After 8 h of stirring at room temperature, the reaction phases were separated, and the aqueous phase was extracted with dichloromethane (3 × 20 mL). The combined organic layer was washed with water, dried over anhydrous magnesium sulfate, filtered, and concentrated in vacuo. The residue was purified by flash chromatography on a silica gel to afford compound 4.

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Table 1. Title Compounds 1a-1q



compd	R <sup>1</sup>	R <sup>2</sup>	compd	R <sup>1</sup>	R <sup>2</sup>
1a	phenvl	MeS	1i	4-t-Bu-phenvl	MeS
1b	phenyl	i-Pr	1k	4-t-Bu-phenyl	i-Pr
1c	2-Cl-phenyl	i-Pr	11	2,4-Cl <sub>2</sub> -phenyl	MeS
1d	4-Cl-phenyl	MeS	1m	2,4-Cl <sub>2</sub> -phenyl	i-Pr
1e	4-Cl-phenyl	i-Pr	1n	i-Pr	MeS
1f	2-OCH <sub>3</sub> -phenyl	MeS	10	i-Pr	i-Pr
1g	2-OCH <sub>3</sub> -phenyl	i-Pr	1p	Br	MeS
1h 1i	4-OCH <sub>3</sub> -phenyl 4-OCH <sub>3</sub> -phenyl	MeS i-Pr	1q	Br	i-Pr

Data for 4a: yield, 73.1%; mp, 87–89 °C (88 °C (14)); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  4.51 (s, 2H), 6.63 (s, 1H), 7.42–7.47 (m, 3H), 7.78–7.81 (m, 2H).

*Data for 4b*: yield, 84.3%; oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  4.53 (s, 2H), 6.80 (s, 1H), 7.32–7.40 (m, 2H), 7.46–7.50 (m, 1H), 7.74 (dd, J = 7.2 Hz, 2.1 Hz, 1H).

Data for 4c: yield, 60.7%; mp, 117–119 °C (118–120 °C (*I5*)); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  4.51 (s, 2H), 6.61 (s, 1H), 7.44 (d, J = 8.4 Hz, 2H), 7.73 (d, J = 8.4 Hz, 2H).

*Data for* **4***d*: yield, 95.1%; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  3.90 (s, 3H), 4.52 (s, 2H), 6.84 (s, 1H), 6.98–7.08 (m, 2H), 7.43 (td, J = 7.8 Hz, 1.8 Hz, 1H), 7.89 (dd, J = 7.8 Hz, 1.8 Hz, 1H).

*Data for 4e:* yield, 60.0%; mp, 86–87 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  3.86 (s, 3H), 4.50 (s, 2H), 5.58 (s, 1H), 6.97 (d, J = 8.7 Hz, 2H), 7.73 (d, J = 8.7 Hz, 2H).

*Data for 4f*: yield, 62.0%; mp, 75–76 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.35 (s, 9H), 4.50 (s, 2H), 6.61 (s, 1H), 7.47 (d, J = 8.4 Hz, 2H), 7.72 (d, J = 8.4 Hz, 2H).

*Data for* **4***g*: yield, 90.1%; mp, 57–59 °C (61–62 °C (*12*)); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  4.53 (s, 2H), 6.68 (s, 1H), 7.35 (dd, J = 8.4 Hz, 2.1 Hz, 1H), 7.52 (d, J = 2.1 Hz, 1H), 7.71 (d, J = 8.4 Hz, 1H).

*Data for 4h*: yield, 78.4%; oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.28 (d, J = 6.9 Hz, 6H), 3.01–3.09 (m, 1H), 4.44 (s, 2H), 6.20 (s, 1H).

Synthetic Procedure for 4i (*16*). A solution of dibromoformaloxime (4.1 g, 20 mmol) in dichloromathane (20 mL) was added over 5 h to a stirred solution of propargyl bromide (2.86 g, 24 mmol), potassium bicarbonate (2.5 g, 30 mmol), dichloromethane (100 mL), and water (10 mL). After 2 h, water (50 mL) was added. The organic layer was separated and dried over anhydrous magnesium sulfate, filtered, and concentrated in vacuo. The residue was purified by flash chromatography on a silica gel to afford compound 4i as a colorless oil, yield 50.0%: <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  4.44 (s, 2H), 6.41 (s, 1H).

**General Synthetic Procedures for 5a-5i.** To a solution of **4** (6 mmol) in *N*,*N*-dimethylformamide (10 mL) was added potassium phthalimide (6 mmol) in portions. After the mixture was stirred at room temperature for 5 h, water (50 mL) was added, and the precipitate was collected by filtration and washed with water. After recrystallization from ethanol, N-substituted phthalimide **5** was obtained as a white crystal.

Data for 5a: yield, 98.5%; mp, 162–164 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  5.04 (s, 2H), 6.57 (s, 1H), 7.41–7.43 (m, 3H), 7.74–7.77 (m, 4H), 7.88–7.92 (m, 2H).

*Data for 5b*: yield, 84.0%; mp, 120–121 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  5.06 (s, 2H), 6.74 (s, 1H), 7.26–7.48 (m, 3H), 7.68 (d, J = 7.2 Hz, 1H), 7.74–7.78 (m, 2H), 7.88–7.92 (m, 2H).

*Data for 5c*: yield, 96.3%; mp, 173–175 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  5.03 (s, 2H), 6.54 (s, 1H), 7.39 (d, J = 8.4 Hz, 2H), 7.68 (d, J = 8.4 Hz, 2H), 7.74–7.78 (m, 2H), 7.88–7.91 (m, 2H).

*Data for 5d*: yield, 98.2%; mp, 149–151 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  3.87 (s, 3H), 5.03 (s, 2H), 6.78 (s, 1H), 6.95–7.02 (m, 2H), 7.38 (t, J = 6.8 Hz, 1H), 7.72–7.76 (m, 2H), 7.81 (dd, J = 6.8 Hz, 4.6 Hz, 1H), 7.87–7.90 (m, 2H).

*Data for 5e*: yield, 89.0%; mp, 177–178 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  3.84 (s, 3H), 5.02 (s, 2H), 6.51 (s, 1H), 6.93 (d, J = 8.7 Hz, 2H), 7.69 (d, J = 8.7 Hz, 2H), 7.74–7.78 (m, 2H), 7.88–7.92 (m, 2H).

*Data for 5f*: yield, 97.0%; mp, 205–206 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.33 (s, 9H), 5.03 (s, 2H), 6.54 (s, 1H), 7.44 (d, J = 8.4 Hz, 2H), 7.69 (d, J = 8.4 Hz, 2H), 7.74–7.78 (m, 2H), 7.88–7.92 (m, 2H).

*Data for* 5g: yield, 85.2%; mp, 152–154 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  5.06 (s, 2H), 6.73 (s, 1H), 7.31 (dd, J = 8.4 Hz, 2.0 Hz, 1H), 7.48 (d, J = 2.0 Hz, 1H), 7.64 (d, J = 8.4 Hz, 1H), 7.75–7.78 (m, 2H), 7.80–7.91 (m, 2H).

Data for 5*h*: yield, 86.0%; mp, 109–110 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.24 (d, J = 6.9 Hz, 6H), 2.97–3.06 (m, 1H), 4.95 (s, 2H), 6.11 (s, 1H), 7.74–7.77 (m, 2H), 7.78–7.91 (m, 2H).

*Data for 5i*: yield, 70.0%; mp, 156−157 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 4.99 (s, 2H), 6.36 (s, 1H), 7.73−7.79 (m, 2H), 7.79−7.92 (m, 2H).

**General Synthetic Procedures for 6a-6i.** To a suspension of N-substituted phthalimide 5 (4 mmol) in ethanol (20 mL) was added hydrazine hydrate (50%, 0.48 g, 4.8 mmol). The reaction mixture was refluxed for 5 h and then cooled. The precipitated phthalylhydrazide was filtered off and washed with ethanol, and then the filtrate was concentrated under reduced pressure to give crude 6, which was utilized in the next reaction without further purification.

*Data for* **6a**: yield, 94.3%; mp, 61–62 °C (51–52 °C (*17*)); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.64 (s, 2H), 4.01 (s, 2H), 6.45 (s, 1H), 7.43–7.46 (m, 3H), 7.77–7.80 (m, 2H).

*Data for 6b*: yield, 98.5%; oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.75 (s, 2H), 4.04 (s, 2H), 6.60 (s, 1H), 7.34–7.38 (m, 2H), 7.46–7.50 (m, 1H), 7.70–7.78 (m, 1H).

*Data for* **6***c*: yield, 82.7%; mp, 79–81 °C (81 °C (*18*)); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.58 (s, 2H), 4.02 (s, 2H), 6.43 (s, 1H), 7.42 (d, *J* = 8.4 Hz, 2H), 7.72 (d, *J* = 8.4 Hz, 2H).

*Data for* **6***d*: yield, 95.0%; oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.87 (s, 2H), 3.85 (s, 3H), 3.98 (s, 2H), 6.61 (s, 1H), 6.94–7.02 (m, 2H), 7.37 (t, J = 7.6 Hz, 1H), 7.82 (d, J = 7.6 Hz, 1H).

*Data for* **6***e*: yield, 92.0%; mp, 111–112 °C;<sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.58 (s, 2H), 3.85 (s, 3H), 4.01 (s, 2H), 6.40 (s, 1H), 6.97 (d, J = 8.8 Hz, 2H), 7.73 (d, J = 8.8 Hz, 2H).

*Data for 6f*: yield, 95.7%; mp, 65–67 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.35 (s, 9H), 1.62 (s, 2H), 4.02 (s, 2H), 6.43 (s, 1H), 7.47 (d, J = 8.4 Hz, 2H), 7.73 (d, J = 8.4 Hz, 2H).

*Data for* **6***g*: yield, 97.2%;<sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.59 (s, 2H), 4.05 (s, 2H), 6.60 (s, 1H), 7.33 (dd, J = 8.4 Hz, 2.0 Hz, 1H), 7.50 (d, J = 2.0 Hz, 1H), 7.67 (d, J = 8.4 Hz, 1H).

*Data for 6h*: yield, 97.0%; oil;<sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.27 (d, J = 6.9 Hz, 6H), 1.60 (s, 2H), 3.01–3.08 (m, 1H), 3.93 (s, 2H), 6.01 (s, 1H).

*Data for 6i*: yield, 95.6%; oil;<sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.57 (s, 2H), 3.94 (s, 2H), 6.21 (s, 1H).

General Synthetic Procedures for the Title Compounds 1a-1q. A mixture of 7a (or 7b) (1.35 mmol) and crude 6 (1.45 mmol) in ethanol (20 mL) was refluxed for 2 h and then evaporated under reduced pressure to give crude product. The residue was purified by flash chromatography on a silica gel to afford the title compounds.

Data for 1a: yield, 98.3%; mp, 86-88 °C;<sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.21 (t,  ${}^{3}J_{HH} = 7.2$  Hz, 3H, CH<sub>3</sub>), 2.72 (s, 3H, SCH<sub>3</sub>), 3.58 (q,  ${}^{3}J_{HH} = 7.2$  Hz, 2H, OCH<sub>2</sub>), 3.71 (t,  ${}^{3}J_{HH} = 5.1$  Hz, 2H, CH<sub>2</sub>O), 4.33 (t,  ${}^{3}J_{HH} = 5.1$  Hz, 2H, CO<sub>2</sub>CH<sub>2</sub>), 4.97 (d,  ${}^{3}J_{HH} = 6.0$  Hz, 2H, CH<sub>2</sub>N), 6.51 (s, 1H, isoxazole), 7.46–7.48 (m, 3H, Ph), 7.78–7.80 (m, 2H, Ph), 10.41 (s, 1H, NH). Anal. Calcd for C<sub>19</sub>H<sub>21</sub>N<sub>3</sub>O<sub>4</sub>S: C, 58.90; H, 5.46; N, 10.85. Found: C, 58.88; H, 5.30; N, 10.68.

Data for 1b: yield, 90.5%; mp, 88–89 °C;<sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.22 (t, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 3H, CH<sub>3</sub>), 1.44 (d, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 6H, C(CH<sub>3</sub>)<sub>2</sub>), 3.19–3.22 (m, 1H, CH), 3.58 (q, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 2H, OCH<sub>2</sub>), 3.71 (t, <sup>3</sup>J<sub>HH</sub> = 5.1 Hz, 2H, CH<sub>2</sub>O), 4.31 (t, <sup>3</sup>J<sub>HH</sub> = 5.1 Hz, 2H, CO<sub>2</sub>CH<sub>2</sub>), 4.77 (d, <sup>3</sup>J<sub>HH</sub> = 6.0 Hz, 2H, CH<sub>2</sub>N), 6.53 (s, 1H, isoxazole), 7.47–7.48 (m, 3H, Ph), 7.78–7.80 (m, 2H, Ph), 10.68 (s, 1H, NH). Anal. Calcd for C<sub>21</sub>H<sub>25</sub>N<sub>3</sub>O<sub>4</sub>: C, 65.51; H, 6.37; N, 11.09. Found: C, 65.78; H, 6.57; N, 10.96.

Data for Ic: yield, 87.3%; mp, 64–65 °C;<sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.21 (t, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 3H, CH<sub>3</sub>), 1.45 (d, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 6H, C(CH<sub>3</sub>)<sub>2</sub>), 3.19–3.22 (m, 1H, CH), 3.58 (q, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 2H, OCH<sub>2</sub>), 3.70 (t, <sup>3</sup>J<sub>HH</sub> = 4.8 Hz, 2H, CH<sub>2</sub>O), 4.30 (t, <sup>3</sup>J<sub>HH</sub> = 4.8 Hz, 2H, CO<sub>2</sub>CH<sub>2</sub>), 4.78 (d, <sup>3</sup>J<sub>HH</sub> = 6.4 Hz, 2H, CH<sub>2</sub>N), 6.69 (s, 1H, isoxazole), 7.35–7.44 (m, 2H,

Ph), 7.50 (t,  ${}^{3}J_{HH} =$  7.6 Hz, 1H, Ph), 7.71 (d,  ${}^{3}J_{HH} =$  7.2 Hz, 1H, Ph), 10.68 (s, 1H, NH). Anal. Calcd for C<sub>21</sub>H<sub>24</sub>ClN<sub>3</sub>O<sub>4</sub>: C, 60.36; H, 5.79; N, 10.06. Found: C, 59.96; H, 5.87; N, 10.11.

Data for 1d: yield, 92.1%; mp, 108–109 °C;<sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.22 (t, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 3H, CH<sub>3</sub>), 2.73 (s, 3H, SCH<sub>3</sub>), 3.58 (q, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 2H, OCH<sub>2</sub>), 3.71 (t, <sup>3</sup>J<sub>HH</sub> = 5.1 Hz, 2H, CH<sub>2</sub>O), 4.33 (t, <sup>3</sup>J<sub>HH</sub> = 5.1 Hz, 2H, CO<sub>2</sub>CH<sub>2</sub>), 4.98 (d, <sup>3</sup>J<sub>HH</sub> = 6.0 Hz, 2H, CH<sub>2</sub>N), 6.49 (s, 1H, isoxazole), 7.45 (d, <sup>3</sup>J<sub>HH</sub> = 8.4 Hz, 2H, Ph), 7.72 (d, <sup>3</sup>J<sub>HH</sub> = 8.4 Hz, 2H, Ph), 10.42 (s, 1H, NH). Anal. Calcd for C<sub>19</sub>H<sub>20</sub>ClN<sub>3</sub>O<sub>4</sub>S: C, 54.09; H, 4.78; N, 9.96. Found: C, 54.94; H, 4.81; N, 9.95.

Data for 1e: yield, 85.0%; mp, 80–82 °C;<sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.22 (t, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 3H, CH<sub>3</sub>), 1.44 (d, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 6H, C(CH<sub>3</sub>)<sub>2</sub>), 3.20–3.21 (m, 1H, CH), 3.59 (q, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 2H, OCH<sub>2</sub>), 3.71 (t, <sup>3</sup>J<sub>HH</sub> = 5.1 Hz, 2H, CH<sub>2</sub>O), 4.31 (t, <sup>3</sup>J<sub>HH</sub> = 5.1 Hz, 2H, CO<sub>2</sub>CH<sub>2</sub>), 4.77 (d, <sup>3</sup>J<sub>HH</sub> = 6.0 Hz, 2H, CH<sub>2</sub>N), 6.51 (s, 1H, isoxazole), 7.45 (d, <sup>3</sup>J<sub>HH</sub> = 8.4 Hz,

Scheme 1. General Synthetic Route for Aminomethyl Isoxazoles 6a-6i



Scheme 2. General Synthetic Route for the Title Compounds 1a-1q

2H, Ph), 7.72 (d,  ${}^{3}J_{HH} = 8.4$  Hz, 2H, Ph), 10.68 (s, 1H, NH). Anal. Calcd for C<sub>21</sub>H<sub>24</sub>ClN<sub>3</sub>O<sub>4</sub>: C, 60.36; H, 5.79; N, 10.06. Found: C, 60.48; H, 5.98; N, 10.03.

Data for If: yield, 98.4%; mp, 80–82 °C;<sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.21 (t, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 3H, CH<sub>3</sub>), 2.72 (s, 3H, SCH<sub>3</sub>), 3.58 (q, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 2H, OCH<sub>2</sub>), 3.71 (t, <sup>3</sup>J<sub>HH</sub> = 5.1 Hz, 2H, CH<sub>2</sub>O), 3.91 (s, 3H, OCH<sub>3</sub>), 4.32 (t, <sup>3</sup>J<sub>HH</sub> = 5.1 Hz, 2H, CO<sub>2</sub>CH<sub>2</sub>), 4.96 (d, <sup>3</sup>J<sub>HH</sub> = 6.0 Hz, 2H, CH<sub>2</sub>N), 6.71 (s, 1H, isoxazole), 7.00–7.07 (m, 2H, Ph), 7.44 (t, <sup>3</sup>J<sub>HH</sub> = 7.6 Hz, 1H, Ph), 7.86 (d, <sup>3</sup>J<sub>HH</sub> = 6.8 Hz, 1H, Ph), 10.39 (s, 1H, NH). Anal. Calcd for C<sub>20</sub>H<sub>23</sub>N<sub>3</sub>O<sub>5</sub>S: C, 57.54; H, 5.55; N,10.07. Found: C, 57.35; H, 5.85; N, 9.79.

Data for Ig: yield, 84.4%; oil,<sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.22 (t, <sup>3</sup> $J_{HH}$  = 7.2 Hz, 3H, CH<sub>3</sub>), 1.45 (d, <sup>3</sup> $J_{HH}$  = 7.2 Hz, 6H, C(CH<sub>3</sub>)<sub>2</sub>), 3.18–3.21 (m, 1H, CH), 3.58 (q, <sup>3</sup> $J_{HH}$  = 7.2 Hz, 2H, OCH<sub>2</sub>), 3.71 (t, <sup>3</sup> $J_{HH}$  = 5.1 Hz, 2H, CH<sub>2</sub>O), 3.85 (s, 3H, OCH<sub>3</sub>), 4.30 (t, <sup>3</sup> $J_{HH}$  = 5.1 Hz, 2H, CO<sub>2</sub>CH<sub>2</sub>), 4.75 (d, <sup>3</sup> $J_{HH}$  = 6.0 Hz, 2H, CH<sub>2</sub>N), 6.71 (s, 1H, isoxazole), 7.00–7.07 (m, 2H, Ph), 7.44 (t, <sup>3</sup> $J_{HH}$  = 7.6 Hz, 1H, Ph), 7.85 (d, <sup>3</sup> $J_{HH}$  = 6.8 Hz, 1H, Ph), 10.67 (s, 1H, NH). HRMS for C<sub>22</sub>H<sub>27</sub>N<sub>3</sub>O<sub>5</sub> + Na: 436.1843. Found: 436.1840.

Data for 1h: yield, 98.0%; oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.20 (t, <sup>3</sup>J<sub>HH</sub> = 6.8 Hz, 3H, CH<sub>3</sub>), 2.71 (s, 3H, SCH<sub>3</sub>), 3.57 (q, <sup>3</sup>J<sub>HH</sub> = 6.8 Hz, 2H, OCH<sub>2</sub>), 3.70 (t, <sup>3</sup>J<sub>HH</sub> = 4.8 Hz, 2H, CH<sub>2</sub>O), 4.32 (t, <sup>3</sup>J<sub>HH</sub> = 4.8 Hz, 2H, CO<sub>2</sub>CH<sub>2</sub>), 4.94 (d, <sup>3</sup>J<sub>HH</sub> = 6.0 Hz, 2H, CH<sub>2</sub>N), 6.44 (s, 1H, isoxazole), 6.97 (d, <sup>3</sup>J<sub>HH</sub> = 8.4 Hz, 2H, Ph), 7.72 (d, <sup>3</sup>J<sub>HH</sub> = 8.4 Hz, 2H, Ph), 10.39 (s, 1H, NH). Anal. Calcd for C<sub>20</sub>H<sub>23</sub>N<sub>3</sub>O<sub>5</sub>S: C, 57.54; H, 5.55; N, 10.07. Found: C, 57.75; H, 5.80; N, 10.01.

Data for Ii: yield, 84.1%; oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.21 (t, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 3H, CH<sub>3</sub>), 1.43 (d, <sup>3</sup>J<sub>HH</sub> = 6.8 Hz, 6H, C(CH<sub>3</sub>)<sub>2</sub>), 3.10–3.30 (m, 1H, CH), 3.58 (q, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 2H, OCH<sub>2</sub>), 3.70 (t, <sup>3</sup>J<sub>HH</sub> = 4.8 Hz, 2H, CH<sub>2</sub>O), 3.85 (s, 3H), 4.30 (t, <sup>3</sup>J<sub>HH</sub> = 4.8 Hz, 2H, CO<sub>2</sub>CH<sub>2</sub>), 4.74 (d, <sup>3</sup>J<sub>HH</sub> = 6.0 Hz, 2H, CH<sub>2</sub>N), 6.46 (s, 1H, isoxazole), 6.97 (d, <sup>3</sup>J<sub>HH</sub> = 8.8 Hz, 2H, Ph), 7.72 (d, <sup>3</sup>J<sub>HH</sub> = 8.8 Hz, 2H, Ph), 10.66 (s, 1H, NH). Anal. Calcd for C<sub>22</sub>H<sub>27</sub>N<sub>3</sub>O<sub>5</sub>: C, 63.91; H, 6.58; N, 10.16. Found: C, 63.77; H, 6.59; N, 10.19.

*Data for Ij*: yield, 92.0%; mp, 94–96 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.21 (t, <sup>3</sup>*J*<sub>HH</sub> = 7.2 Hz, 3H, CH<sub>3</sub>), 1.35 (s, 9H, C(CH<sub>3</sub>)<sub>3</sub>), 2.72 (s, 3H, SCH<sub>3</sub>), 3.58 (q, <sup>3</sup>*J*<sub>HH</sub> = 7.2 Hz, 2H, OCH<sub>2</sub>), 3.71 (t, <sup>3</sup>*J*<sub>HH</sub> = 5.1 Hz, 2H, CH<sub>2</sub>O), 4.33



 Table 2.
 Herbicidal Activities of Compounds1a-1q (1.5 kg/ha, Percent Inhibition)

	postemergence treatment				preemergence treatment			
compd	rape	amaranth pigweed	barnyard grass	hairy crabgrass	rape	amaranth pigweed	barnyard grass	hairy crabgrass
1a	100	100	57.0	55.5	15.8	67.7	0	0
1b	100	100	40.0	59.4	28.7	88.6	0	24.9
1c	100	100	47.2	24.0		0	0	0
1d	100	89.9	58.2	72.7	28.4	23.1	0	14.8
1e	100	100	33.0	64.1	10.6	68.7	0	0
1f	100	90.6	52.0	52.3	19.0	47.8	0	17.3
1g	100	95.1	56.4	82.0	17.4	44.9	0	5.5
1h	100	100	0	32.5	10.0	0	0	0
1i	100	100	0	14.3	40.9	45.2	0	0
1j	100	100	0	55.8	0	10.0	0	0
1k	100	100	0	24.7	0	0	0	0
11	100	89.9	45.2	64.1	45.0	65.8	0	15.6
1m	100	100	56.7	63.3	21.3	50.2	0	18.1
1n	100	94.0	32.6	48.4		25.0	0	0
10	100	100	12.5	48.4		100	5.0	0
1p	100	99.1	36.8	55.3		0	2.4	0
1q	100	100	66.5	42.1		60.0	22.6	0
Α	100	100	55.3	54.1	47.6	32.4	3.6	18.8

(t,  ${}^{3}J_{HH} = 5.1 \text{ Hz}, 2\text{H}, \text{CO}_2\text{CH}_2$ ),  $4.97 \text{ (d}, {}^{3}J_{HH} = 6.0 \text{ Hz}, 2\text{H}, \text{CH}_2\text{N}$ ), 6.48 (s, 1H, isoxazole),  $7.27 \text{ (d}, {}^{3}J_{HH} = 8.4 \text{ Hz}, 2\text{H}, \text{Ph}$ ),  $7.48 \text{ (d}, {}^{3}J_{HH} = 8.4 \text{ Hz}, 2\text{H}, \text{Ph}$ ), 10.40 (s, 1H, NH). Anal. Calcd for  $\text{C}_{23}\text{H}_{29}\text{N}_3\text{O}_4\text{S}$ : C, 62.28; H, 6.59; N, 9.47. Found: C, 62.18; H, 6.56; N, 9.50.

*Data for Ik*: yield, 51.0%; oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.21 (t, <sup>3</sup> $J_{HH}$  = 6.9 Hz, 3H, CH<sub>3</sub>), 1.35 (s, 9H, C(CH<sub>3</sub>)<sub>3</sub>), 1.44 (d, <sup>3</sup> $J_{HH}$  = 7.2 Hz, 6H, C(CH<sub>3</sub>)<sub>2</sub>), 3.14–3.26 (m, 1H, CH), 3.58 (q, <sup>3</sup> $J_{HH}$  = 6.9 Hz, 2H, OCH<sub>2</sub>), 3.71 (t, <sup>3</sup> $J_{HH}$  = 5.1 Hz, 2H, CH<sub>2</sub>O), 4.31 (t, <sup>3</sup> $J_{HH}$  = 5.1 Hz, 2H, CO<sub>2</sub>CH<sub>2</sub>), 4.75 (d, <sup>3</sup> $J_{HH}$  = 6.0 Hz, 2H, CH<sub>2</sub>N), 6.50 (s, 1H, isoxazole), 7.48 (d, <sup>3</sup> $J_{HH}$  = 8.4 Hz, 2H, Ph), 7.72 (d, <sup>3</sup> $J_{HH}$  = 8.4 Hz, 2H, Ph), 10.67 (s, 1H, NH). HRMS for C<sub>25</sub>H<sub>33</sub>-N<sub>3</sub>O<sub>4</sub> + Na: 462.2363. Found: 462.2357.

Data for 11: yield, 91.2%; mp, 85–86 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.21 (t, <sup>3</sup>J<sub>HH</sub> = 6.9 Hz, 3H, CH<sub>3</sub>), 2.73 (s, 3H, SCH<sub>3</sub>), 3.58 (q, <sup>3</sup>J<sub>HH</sub> = 6.9 Hz, 2H, OCH<sub>2</sub>), 3.71 (t, <sup>3</sup>J<sub>HH</sub> = 5.1 Hz, 2H, CH<sub>2</sub>O), 4.33 (t, <sup>3</sup>J<sub>HH</sub> = 5.1 Hz, 2H, CO<sub>2</sub>CH<sub>2</sub>), 5.00 (d, <sup>3</sup>J<sub>HH</sub> = 6.0 Hz, 2H, CH<sub>2</sub>N), 6.68 (s, 1H, isoxazole), 7.34–7.36 (m, 1H, Ph), 7.50–7.54 (m, 1H, Ph), 7.66–7.69 (m, 1H, Ph), 10.41 (s, 1H, NH). Anal. Calcd for C<sub>19</sub>H<sub>19</sub>Cl<sub>2</sub>N<sub>3</sub>O<sub>4</sub>S: C, 50.01; H, 4.20; N, 9.21. Found: C, 50.08; H, 4.31; N, 9.16.

Data for 1m: yield, 70.6%; oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.21 (t, <sup>3</sup>J<sub>HH</sub> = 6.9 Hz, 3H, CH<sub>3</sub>), 1.45 (d, <sup>3</sup>J<sub>HH</sub> = 6.9 Hz, 6H, C(CH<sub>3</sub>)<sub>2</sub>), 3.17–3.23 (m, 1H, CH), 3.58 (q, <sup>3</sup>J<sub>HH</sub> = 6.9 Hz, 2H, OCH<sub>2</sub>), 3.71 (t, <sup>3</sup>J<sub>HH</sub> = 5.1 Hz, 2H, CH<sub>2</sub>O), 4.31 (t, <sup>3</sup>J<sub>HH</sub> = 5.1 Hz, 2H, CO<sub>2</sub>CH<sub>2</sub>), 4.79 (d, <sup>3</sup>J<sub>HH</sub> = 6.0 Hz, 2H, CH<sub>2</sub>N), 6.68 (s, 1H, isoxazole), 7.34–7.37 (m, 1H, Ph), 7.51–7.55 (m, 1H, Ph), 7.65–7.68 (m, 1H, Ph), 10.68 (s, 1H, NH). HRMS for C<sub>21</sub>H<sub>23</sub>Cl<sub>2</sub>N<sub>3</sub>O<sub>4</sub> + Na: 474.0958. Found: 474.0954.

*Data for In*: yield, 90.6%; mp, 49–50 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.21 (t, <sup>3</sup>*J*<sub>HH</sub> = 7.2 Hz, 3H, CH<sub>3</sub>), 1.28 (d, <sup>3</sup>*J*<sub>HH</sub> = 7.2 Hz, 6H, C(CH<sub>3</sub>)<sub>2</sub>), 2.70 (s, 3H, SCH<sub>3</sub>), 3.02–3.09 (m, 1H, CH), 3.58 (q, <sup>3</sup>*J*<sub>HH</sub> = 7.2 Hz, 2H, OCH<sub>2</sub>), 3.71 (t, <sup>3</sup>*J*<sub>HH</sub> = 4.8 Hz, 2H, CH<sub>2</sub>O), 4.32 (t, <sup>3</sup>*J*<sub>HH</sub> = 4.8 Hz, 2H, CO<sub>2</sub>CH<sub>2</sub>), 4.88 (d, <sup>3</sup>*J*<sub>HH</sub> = 6.0 Hz, 2H, CH<sub>2</sub>O), 6.06 (s, 1H, isoxazole), 10.34 (s, 1H, NH). Anal. Calcd for C<sub>16</sub>H<sub>23</sub>N<sub>3</sub>O<sub>4</sub>S: C, 54.37; H, 6.56; N, 11.89. Found: C, 54.23; H, 6.54; N, 11.86.

Data for Io: yield, 89.5%; oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 1.22 (t, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 3H, CH<sub>3</sub>), 1.28 (d, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 6H, C(CH<sub>3</sub>)<sub>2</sub>), 1.42 (d, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 6H, C(CH<sub>3</sub>)<sub>2</sub>), 3.01-3.11 (m, 1H, CH), 3.12-3.22 (m, 1H, CH), 3.58 (q, <sup>3</sup>J<sub>HH</sub> = 7.2 Hz, 2H, OCH<sub>2</sub>), 3.71 (t, <sup>3</sup>J<sub>HH</sub> = 4.8 Hz, 2H, CH<sub>2</sub>O), 4.30 (t, <sup>3</sup>J<sub>HH</sub> = 4.8 Hz, 2H, CO<sub>2</sub>CH<sub>2</sub>), 4.67 (d, <sup>3</sup>J<sub>HH</sub> = 6.0 Hz, 2H, CH<sub>2</sub>N), 6.07 (s, 1H, isoxazole), 10.61 (s, 1H, NH). HRMS for C<sub>18</sub>H<sub>27</sub>N<sub>3</sub>O<sub>4</sub> + Na: 372.1894. Found: 372.1887.

Data for Ip: yield, 87.6%; oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.21 (t, <sup>3</sup> $J_{HH}$  = 6.9 Hz, 3H, CH<sub>3</sub>), 2.71 (s, 3H, SCH<sub>3</sub>), 3.57 (t, <sup>3</sup> $J_{HH}$  = 6.9 Hz, 2H, OCH<sub>2</sub>), 3.70 (t, <sup>3</sup> $J_{HH}$  = 5.1 Hz, 2H, CH<sub>2</sub>O), 4.32 (t, <sup>3</sup> $J_{HH}$  = 5.1 Hz, 2H, CO<sub>2</sub>CH<sub>2</sub>), 4.92 (d, <sup>3</sup> $J_{HH}$  = 6.3 Hz, 2H, CH<sub>2</sub>N), 6.30 (s, 1H, isoxazole), 10.34 (s, 1H, NH). HRMS for C<sub>13</sub>H<sub>16</sub>BrN<sub>3</sub>O<sub>4</sub>S + Na: 411.9937. Found: 411.9935.

Data for Iq: yield, 80.0%; oil; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.21 (t, <sup>3</sup> $J_{HH}$  = 7.2 Hz, 3H, CH<sub>3</sub>), 1.42 (d, <sup>3</sup> $J_{HH}$  = 7.2 Hz, 6H, C(CH<sub>3</sub>)<sub>2</sub>), 3.10–3.20 (br s, 1H, CH), 3.58 (q, <sup>3</sup> $J_{HH}$  = 7.2 Hz, 2H, OCH<sub>2</sub>), 3.70 (t, <sup>3</sup> $J_{HH}$  = 4.8 Hz, 2H, CH<sub>2</sub>O), 4.30 (t, <sup>3</sup> $J_{HH}$  = 4.8 Hz, 2H, CO<sub>2</sub>CH<sub>2</sub>), 4.74 (d, <sup>3</sup> $J_{HH}$  = 6.0 Hz, 2H, CH<sub>2</sub>N), 6.34 (s, 1H, isoxazole), 10.61 (s, 1H, NH). HRMS for C<sub>15</sub>H<sub>20</sub>BrN<sub>3</sub>O<sub>4</sub> + Na: 408.0529. Found: 408.0525.

Herbicidal Activity. Two dicotyledon species, rape (*Brassica napus* L.) and amaranth pigweed (*Amaranthus retroflexus*), and two monocotyledon weeds, barnyard grass (*Echinochloa crusgalli* (L.) Beauv) and hairy crabgrass (*Digitaria sanguinalis* L. Scop.), were used to test the herbicidal activities of compounds 1a-1q using a previously reported procedure (2).

#### **RESULTS AND DISCUSSION**

Synthesis. Oximes 3a-3h were prepared from corresponding aldehydes (2a-2h) according to a classical procedure. In the presence of sodium hypochlorite, compounds 3a-3h first generated nitrile oxide and subsequently reacted with propargyl bromide to give isoxazoles 4a-4h via 1,3-dipolar cycloaddition. 4i was similarly prepared by reacting dibromoformaloxime with propargyl bromide in the presence of potassium bicarbonate. Bromomethyl compounds 4a-4i were converted to corresponding aminomethyl isoxazoles 6a-6i by Gabriel reaction (Scheme 1).

Alkyoxy- or methylthio-substituted cyanoacrylate 7a and 7b were prepared according to our previous work (2, 4). The title

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Table 3. Herbicidal Activities of Compounds 1a-1q in Postemergence Treatment (Percent Inhibition)

compd	dose (kg/ha)	rape	amaranth pigweed
1a	750	100	34.0
	375	17.8	7.0
1b	750	100	100
	375	100	100
	150	97.3	78.3
	75	54.2	17.4
1c	750	100	98.3
	375	100	96.0
	150	100	94.4
	75	100	77.0
1d	750	0	22.0
	375	0	4.0
1e	750	100	100
	375	100	100
	150	36.3	26.1
	75	29.6	87
1f	750	43.8	16.0
	375	15.8	0
10	750	10.0	100
ig	750	100	100
	375	07.0	100
	130	27.0	23.9
41.	75	17.1	17.4
1 <b>n</b>	/50	94.6	87.0
	375	57.5	34.8
11	750	100	100
	375	100	100
	150	100	93.5
	75	82.5	65.2
1j	750	22.2	45.7
	375	9.8	28.3
1k	750	85.8	89.1
	375	41.7	54.3
11	750	0	46.0
	375	0	0
1m	750	100	100
	375	100	100
	150	21.8	19.6
	75	11.4	10.9
1n	750	76.9	37.6
	375	52.3	0
10	750	100	100
	375	97.7	54.6
	150	92.8	0
	75	52.0	0
1p	750	93.0	0
.6	375	87.0	0
10	750	100	100
• ୩	375	100	100
	150	100	
	75	100	76 F
٨	75	100	40.J 97 0
~	100	100	01.3
	3/3	100	10.7
	150	100	U
	/5	80.9	U

compounds 1a-1q were synthesized from 6 and 7 with good yields (Scheme 2 and Table 1).

Herbicidal Activity Bioassay. Herbicidal activities of the title compounds 1a-1q and control compound A, a previously prepared pyridiyl analogue, are listed in Table 2. Like other 2-cyanoacrylates in our previous work (2–7), all of the title compounds showed higher herbicidal activities in postemergence treatment as compared to preemergence treatment. To our pleasure, most of the compounds exhibited 100% inhibition against dicotyledon species (rape and amaranth pigweed) at 1.5 kg/ha, and half of them showed >50% inhibition against

monocotyledon weeds (barnyard grass and hairy crabgrass), which indicated that the introduction of the isoxazole ring to 2-cyanoarylate effectively enhanced their herbicidal activities.

Their herbicidal activities at lower doses against rape and amaranth pigweed in postemergence treatment revealed the influence of substituents on their reactivity (Table 3). 3-Isopropylacrylate compounds exhibited much higher activities than 3-methylthio analogues; whereas 1b, 1c, 1e, 1g, 1i, 1m, and 1g maintained 100% inhibition at a dose of 375 g/ha, most of 3-methylthio analogues gave only 20% control at the same dose. When substituents at the 3-position of the isoxazole were tested, bromo-substituted compounds (1p and 1q) gave better herbicidal activities than isopropyl and phenyl analogues (1n, 1o, 1a, and **1b**), which can be explained as higher electron negativity. The activities of substituted-phenyl-containing compounds varied largely with the groups on the phenyl group. For instance, from data of herbicidal activities against rape, the percent inhibition of 1c ( $\mathbf{R}^1 = 2$ -Cl-phenyl) was higher than that of 1b ( $\mathbf{R}^1 = phenyl$ ) and  $1e (R^1 = 4$ -Cl-phenyl), whereas  $1g (R^1 = 2$ -OCH<sub>3</sub>-phenyl) was showed less inhibition than  $1b (R^1 = phenyl)$  and  $1i (R^1 = phenyl)$ 4-OCH<sub>3</sub>-phenyl). Therefore, both the electrostatic and steric effects dominated their activities, but 4-tert-butylphenyl compounds (1k) gave the least herbicidal activity, which indicated a bulky group at the para position of phenyl to be not suitable.

Among all of the compounds,  $\mathbf{1c} (\mathbf{R}^1 = 2\text{-}Cl\text{-}phenyl, \mathbf{R}^2 = i\text{-}Pr)$ ,  $\mathbf{1i} (\mathbf{R}^1 = 4\text{-}OCH_3\text{-}phenyl, \mathbf{R}^2 = i\text{-}Pr)$ , and  $\mathbf{1q} (\mathbf{R}^1 = \mathbf{Br}, \mathbf{R}^2 = i\text{-}Pr)$  exhibited excellent activities against rape and good activities against amaranth pigweed even at a dose of 75 g/ha, which was better than compound **A**. However, the activities of their methylthio analogues (**1a** and **1p**) were still lower than that of compound **A**. Comprehensive QSAR will be carried out with other types of compounds.

In summary, 2-cyanoacrylates containing an isoxazole moiety were synthesized and their herbicidal activities against four species were evaluated. Most of the compounds exhibited good inhibition against dicotyledon species (rape and amaranth pigweed) in postemergence treatment, of which 3-isopropylacrylate compounds gave much higher activities than 3-methylthio analogues. The substituents on the 3-position of the isoxazole ring varied the activities to some extent. Compounds **1c**, **1i**, and **1q** exhibited excellent herbicidal activities against rape and good activities against amaranth pigweed even at a dose of 75 g/ha. These compounds deserved further investigation, which we will report in the future.

### LITERATURE CITED

- (1) Huppatz, J. L.; McFadden, H. G.; Huber, M.-L.; McCaffery, L. F. Cyanoacrylate inhibitors of photosynthetic electron transport. Structural requirements for inhibitor potency and herbicidal activity. In *Synthesis and Chemistry of Agrochemicals III*; Baker, D. R., Fenyes, J. G., Steffens, J. J., Eds.; Maple Press: New York, 1992; pp 186–199.
- (2) Wang, Q. M.; Sun, H. K.; Cao, H. Y.; Cheng, M. R.; Huang, R. Q. Synthesis and herbicidal activity of 2-cyano-3-substitutedpyridinemethylaminoacrylates. J. Agric. Food Chem. 2003, 51, 5030–5035.
- (3) Liu, Y. X.; Zhao, Q. Q.; Wang, Q. M.; Li, H.; Huang, R. Q.; Li, Y. H. Synthesis and herbicidal activity of 2-cyano-3-(2-fluoro-5-pyridyl)methylaminoacrylates. J. Fluorine Chem. 2005, 126, 345–348.

- (4) Wang, Q. M.; Li, H.; Cao, H. Y.; Li, Y. H.; Huang, R. Q. Synthesis and herbicidal activity of 2-cyano-3-(2-chlorothiazol-5-yl)methylaminoacrylates. J. Agric. Food Chem. 2004, 52, 1918–1922.
- (5) Liu, Y. X; Cai, B. L.; Li, Y. H.; Song, H. B.; Huang, R. Q.; Wang, Q. M. Synthesis, crystal structure, and biological activities of 2-cyanoacrylates containing furan or tetrahydrofuran moieties. *J. Agric. Food Chem.* **2007**, *55*, 3011–3017.
- (6) Liu, Y. X.; Wei, D. G.; Zhu, Y. R.; Liu, S. H.; Zhang, Y. L.; Zhao, Q. Q.; Cai, B. L.; Li, Y. H.; Song, H. B.; Liu, Y.; Wang, Y.; Huang, R. Q.; Wang, Q. M. Synthesis, herbicidal activities, and 3D-QSAR of 2-cyanoacrylates containing aromatic methylamine moieties. J. Agric. Food Chem. 2008, 56, 204–212.
- (7) Han, X. F.; Liu, Y. X.; Liu, Y.; Lai, L. H.; Huang, R. Q.; Wang, Q. M. Binding model and 3D-QSAR of 3-(2-chloropyrid-5-ylmethylamino)-2-cyanoacrylates as PSII electron transport inhibitor. *Chin. J. Chem.* 2007, 25, 1135–1138.
- (8) Zhao, Q. Q.; Liu, S. H.; Li, Y. H.; Wang, Q. M. Design, synthesis, and biological activities of novel 2-cyanoacrylates containing oxazole, oxadiazole, or quinoline moieties. J. Agric. Food Chem. 2009, 57, 2849–2855.
- (9) Song, B. A.; Yang, S.; Hong, Y. P.; Zhang, G. P.; Jin, L. H.; Hu, D. Y. Synthesis and bioactivity of fluorine compounds containing isoxazolylamino and phosphonate groups. *J. Fluorine Chem.* 2005, *126*, 1419–1424.
- (10) Sperry, J.; Wright, D. Furans, thiophenes and related heterocycles in drug discovery. *Curr. Opin. Drug Discov. Dev.* 2005, *8*, 723–740.
- (11) Kelly, D. R.; Baker, S. C.; King, D. S.; de Silva, D. S.; Lord, G.; Taylor, J. P. Studies of nitrile oxide cycloadditions, and the phenolic oxidative coupling of vanillin aldoxime by *Geobacillus* sp. DDS012 from Italian rye grass silage. *Org. Biomol. Chem.* 2008, 6 (4), 787–796.
- (12) Wade, P. A.; Bereznak, J. F.; Palfey, B. A.; Carroll, P. J.; Dailey, W. P.; Sivasubramanian, S. Diastereofacial selectivity studies on 3-alkenyl-4,5-diphenyl-4,5-dihydroisoxazoles. J. Org. Chem. 1990, 55, 3045–3051.
- (13) Hatta, T.; Kawano, M.; Maeda, H.; Tsuge, O. Synthesis of dithiaand tetrathiacyclophanes incorporating isoxazole units. J. Heterocycl. Chem. 1997, 34 (2), 579–583.
- (14) Bianchi, G.; Grünanger, P. Conversion of 2-isoxazolines to isoxazoles. *Tetrahedron* **1965**, *21*, 817–822.
- (15) Sen, H. G.; Seth, D.; Joshi, U. N.; Rajagopalan, P. 3-Aryl-5halomethylisoxazoles. A new class of anthelminitics. J. Med. Chem. 1966, 9 (3), 431–433.
- (16) Chiarino, D.; Napoletano, M.; Sala, A. One pot synthesis of 3-chloro-5-substituted isoxazoles by 1,3-dipolar cycloaddition. *Synth. Commun.* **1988**, *18* (10), 1171–1176.
- (17) Pei, Y.; Wickham, B. O. S. Regioselective syntheses of 3-aminomethyl-5-substituted isoxazoles: A facile and chemoselective reduction of azide to amine by sodium borohydride using 1,3propanedithiol as a catalyst. *Tetrahedron Lett.* **1993**, *34* (47), 7509– 7512.
- (18) Davenport, J. D.; Dreikorn, B. A.; Elsasser, A. F. Fungicides Mittel. DE 2723688, 1977.

Received for review July 22, 2009. Revised manuscript received September 28, 2009. Accepted November 19, 2009. We gratefully acknowledge the National Key Project for Basic Research (2010CB126100) and the National Natural Science Foundation of China (20972080) for financial support.